

# Climate Change Impacts on Primary Production and Biocalcification

Marion Gehlen

Laboratoire des Sciences du Climat et de l'Environnement

Institut Pierre-Simon Laplace

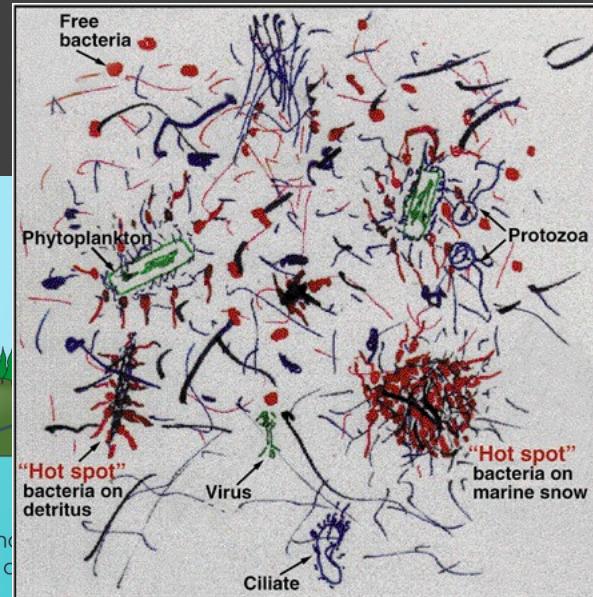
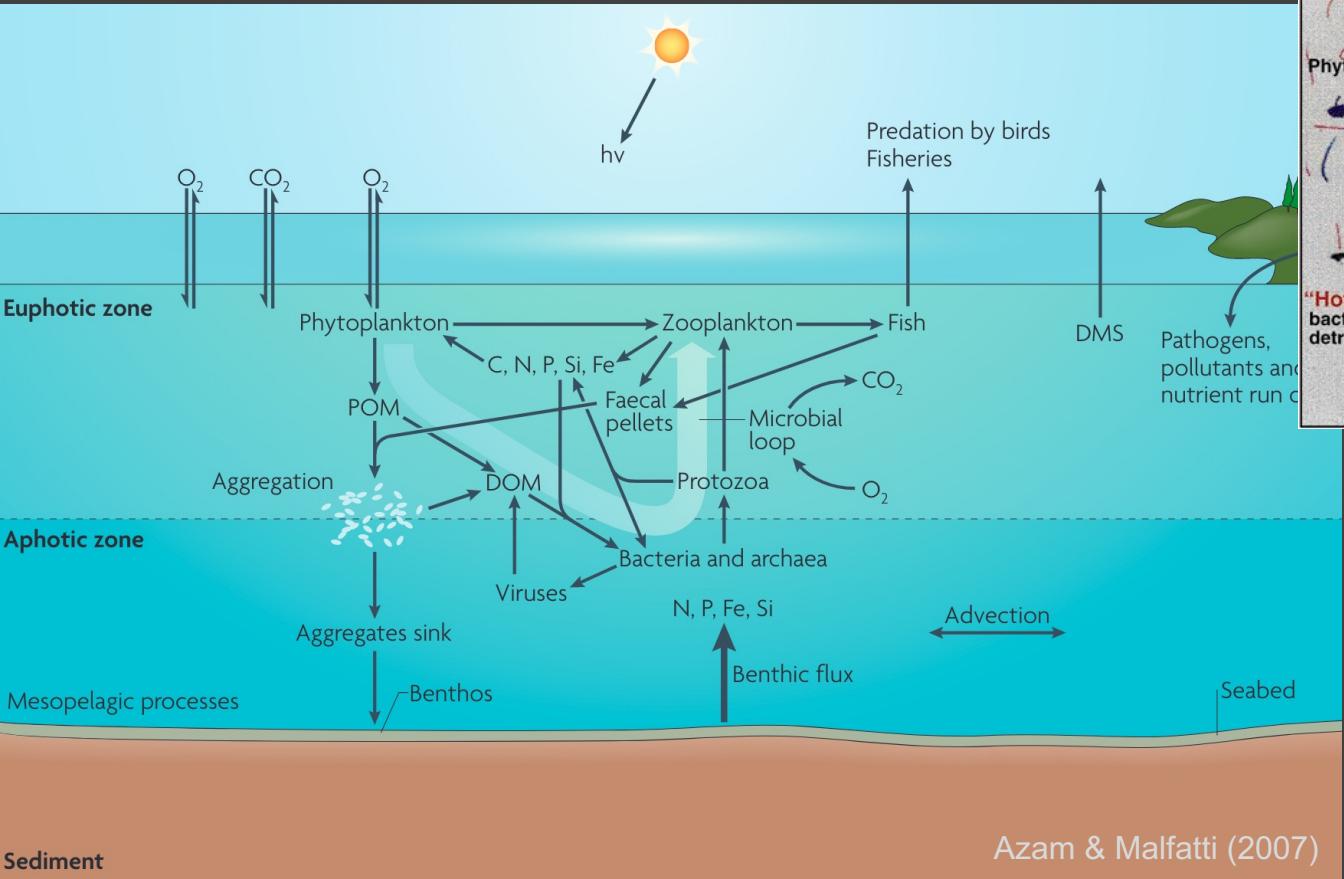


---

LABORATOIRE DES SCIENCES DU CLIMAT & DE L'ENVIRONNEMENT



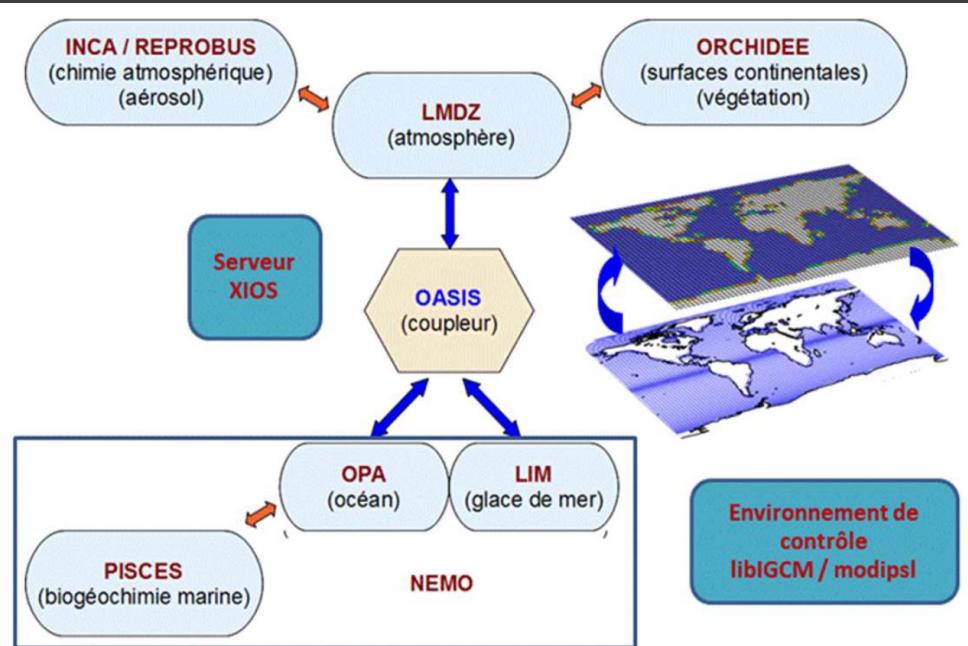
# A microbiologist's view of the ocean ecosystem

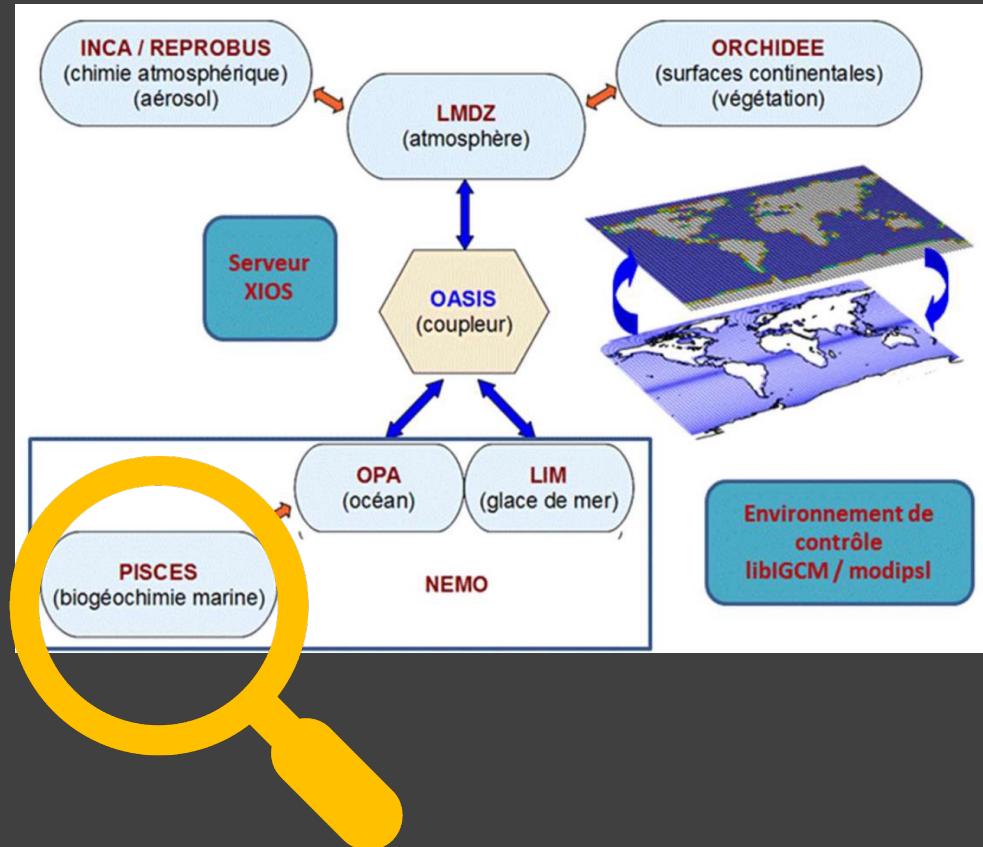


Azam (1998)

Azam & Malfatti (2007)

# IPSL CM6





IPSL CM6

A Biogeochemist's view of the  
ocean ecosystem

# Model of intermediate complexity

24 tracers

4 PFTs

2 phytoplankton

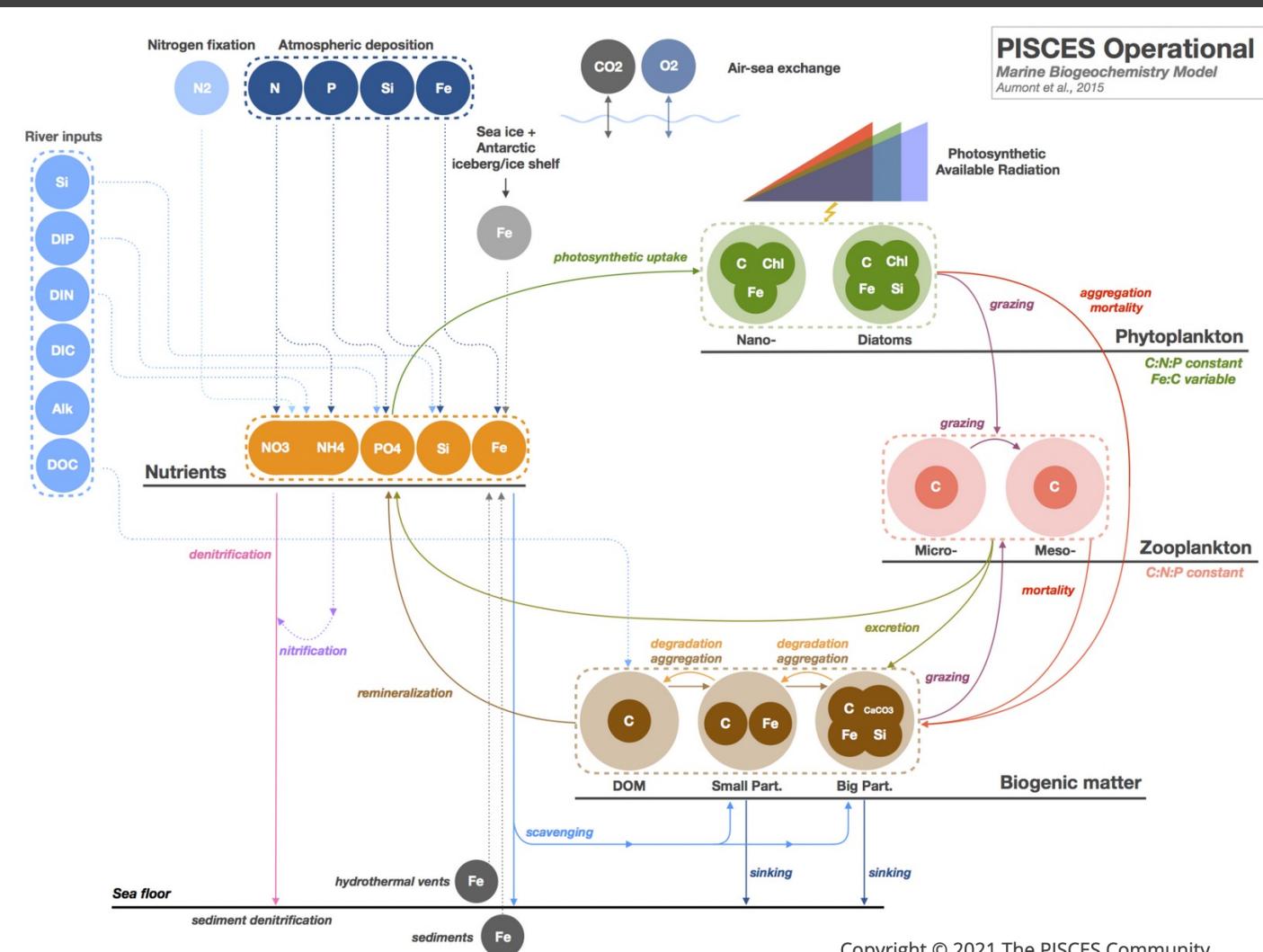
2 zooplankton

+ grazing

+ sinking

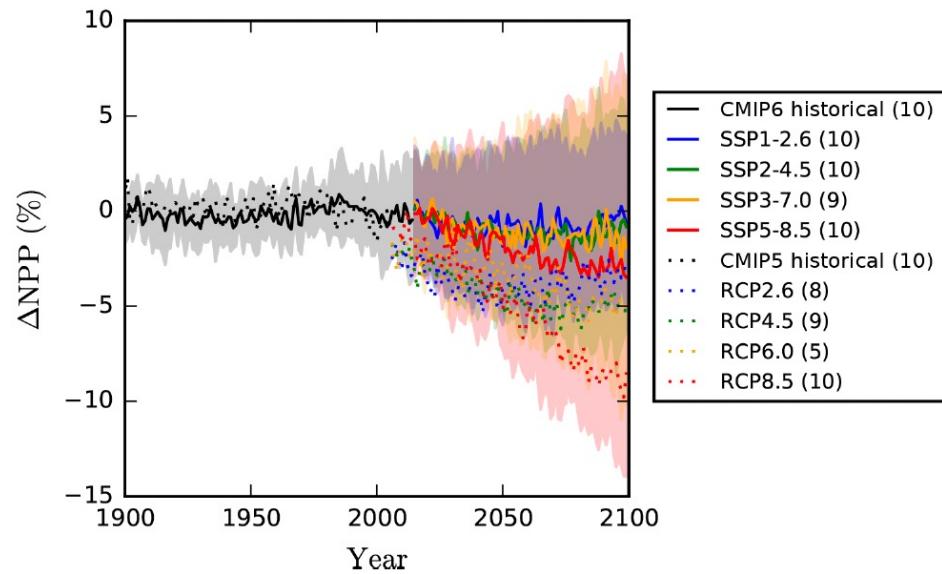
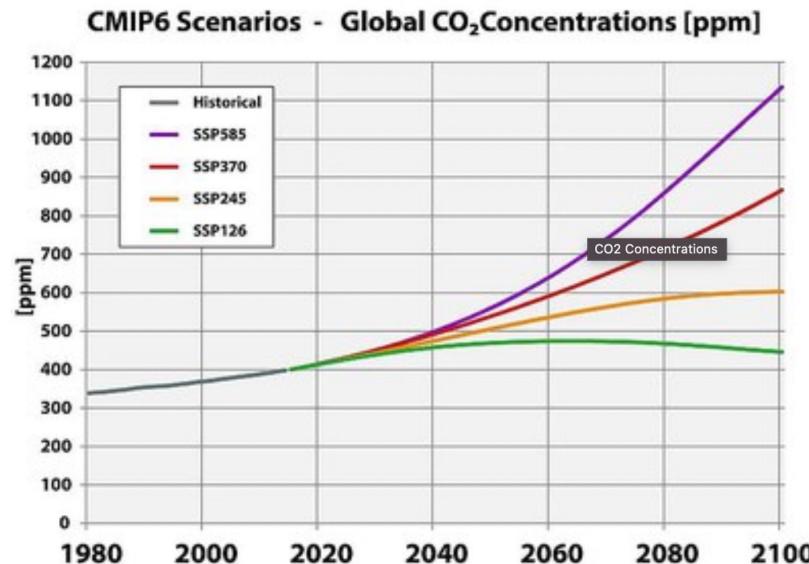
+ implizit representation  
of microbial loop

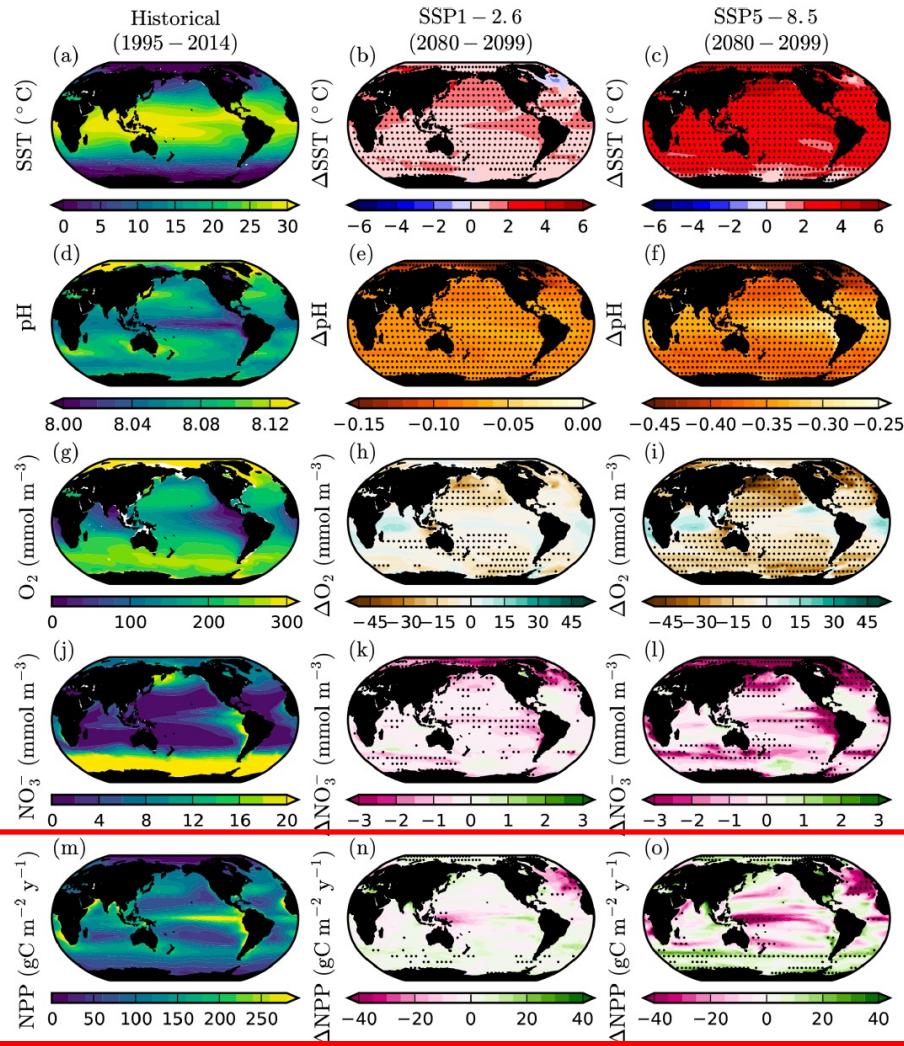
+ implizit representation  
of  $\text{CaCO}_3$  production



One PFT is described by a single set of parameters (growth, nutrient uptake affinity, mortality etc.), but modulated by environmental conditions

# Net Primary Production: Future Evolution in response to climate change





# Drivers of upper-ocean ecosystem impacts

Anomalies are 2080–2099 mean values relative to the 1995–2014 baseline period

sea surface temperature (°C)

surface ocean pH

subsurface dissolved  $O_2$  concentration  
(mean 100 – 600m)

euphotic-zone  $\text{NO}_3^-$  (mean 0 -100 m)

depth integrated NPP

One PFT is described by a single set of parameters (growth, nutrient uptake affinity, mortality etc.), but modulated by environmental conditions

- no plasticity of response to changing climate (acclimation, adaptation)
- no representation of biodiversity (more PFTs ≠ diversity)
- no or little functional redundancy (beyond PP, grazing)

=> function might be unrealistically down(up) regulated under future climate change

## Carbonate Production : standard model, calcite

nutrient limitation

$$\frac{PIC}{POC} = \frac{PIC^*}{POC} L_{\text{lim}}^P \max\left(0.0001, \frac{\text{temp}}{2 + \text{temp}}\right) \max\left(1, \frac{NANOPHYTO}{2}\right)$$

biomass criterion

temperature effect

$$\frac{PIC^*}{POC} = \left(\frac{PIC}{POC}\right)_{\text{max}} \frac{(\Omega_c - 1)}{K_{\text{max}} + (\Omega_c - 1)}$$

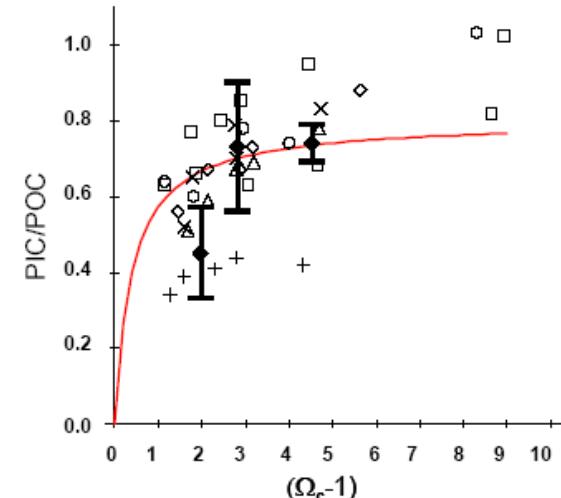
where:

$$\Omega_c = \frac{[\text{Ca}^{2+}]_{\text{SW}} \times [\text{CO}_3^{2-}]_{\text{SW}}}{K_{\text{sp}}^*}$$

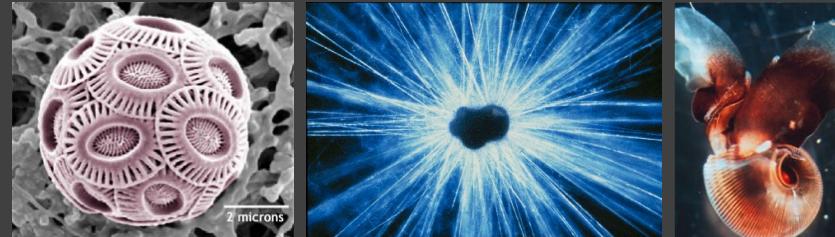
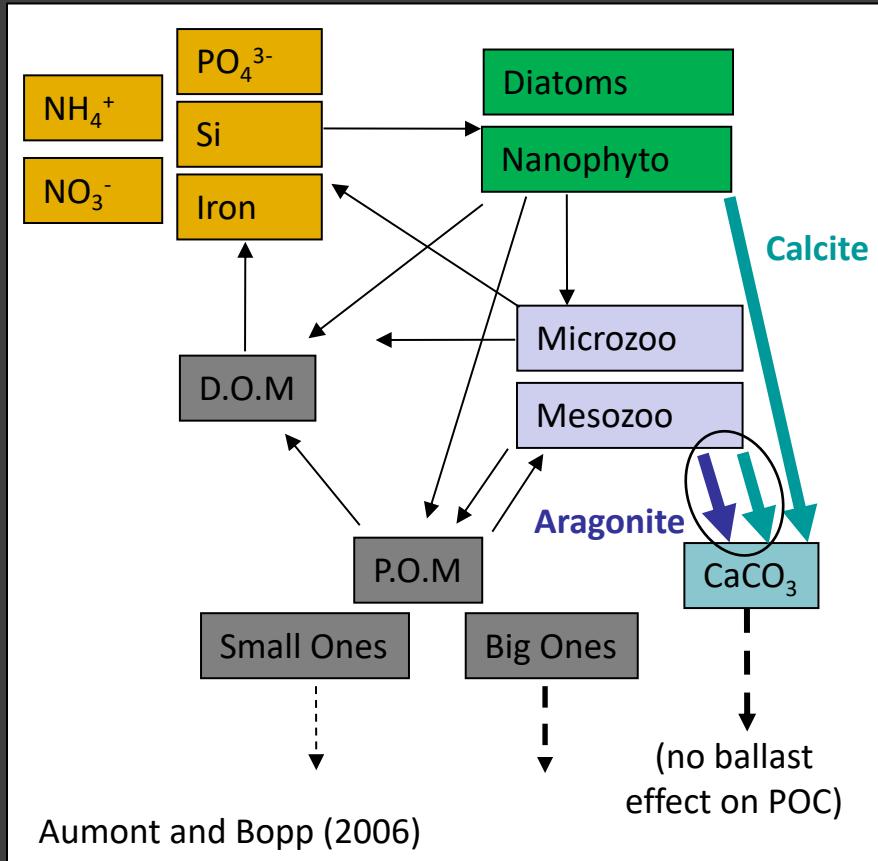
$$K_{\text{sp}}^* = [\text{Ca}^{2+}]_{\text{SAT}} \times [\text{CO}_3^{2-}]_{\text{SAT}}$$

calcite stoichiometric solubility product

a function of pressure, salinity and temperature (Mucci, 1983)



# Carbonate Production : Calcite & Aragonite



CaCO<sub>3</sub> production is attributed to:

1. **Nanophytoplankton**  
= Calcite
2. **Mesozooplankton**  
= Aragonite (new)
3. **Mesozooplankton**  
= Calcite (new)

## Maximum feedback on atm. CO<sub>2</sub> from multiple model simulations

Year	Max feedback on atm. CO <sub>2</sub>	CO <sub>2</sub> perturbations
1766-2100	11.4 ppm	~980 ppm
1766-2500	24 ppm	~600 ppm

⇒ Feedbacks on atmospheric CO<sub>2</sub> from decreasing CaCO<sub>3</sub> production are small

## Exploration of sensitivity of marine carbonate cycle to climate change ocean acidification:

- implicit representation of multiple calcifiers
  - quantification of feedbacks to atm CO<sub>2</sub> robust across a range of models (no shown)
  - decrease of CaCO<sub>3</sub> production highest with zooplankton calcifiers
- => projections of biogeochemical function « CaCO<sub>3</sub> production »
- possible with current ESMs, but no ecosystem impacts

Thank you for your attention !

